

High-Frequency and -Field Electron Paramagnetic Resonance Studies of Yellow/ Orange/ Red Pigments Based on Iron³⁺

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The starting material of this work was YInO_3 , which is amenable to doping of its trigonal bipyramidal (TBP) indium³⁺ site by a wide range of metal ions. Among these are manganese³⁺ (spin number $S = 2$) and iron³⁺ ($S = 5/2$). The former, $\text{YIn}_{1-x}\text{Mn}_x\text{O}_3$, was the subject of an earlier investigation at the MagLab (Ref. 1) and has been successfully commercialized as a novel and attractive blue pigment in a variety of applications including paints and heat-reflective roof covers. Unlike manganese³⁺ which produces a bright blue color, iron³⁺ ion imparts a color on the otherwise colorless (off-white) indium compound varying from pale yellow to orange to intense red depending on its concentration (see Fig. 1).

The $\text{YIn}_{1-x}\text{Fe}_x\text{O}_3$ ($x = 0.1$ to 0.3) series of materials was synthesized in the group of Prof. Subramanian at Oregon State University. Since the optical properties (color) of the materials depend on the electronic structure of iron³⁺, which is paramagnetic, these properties are correlated with the magnetic properties of that ion. For that reason, they were studied by EPR, and in particular its high-frequency and -field version (HFEPR) at the National MagLab in Tallahassee.

Like their manganese-based analogs, the current class of iron-based pigments offers a non-toxic alternative to more conventional pigments in a variety of applications.

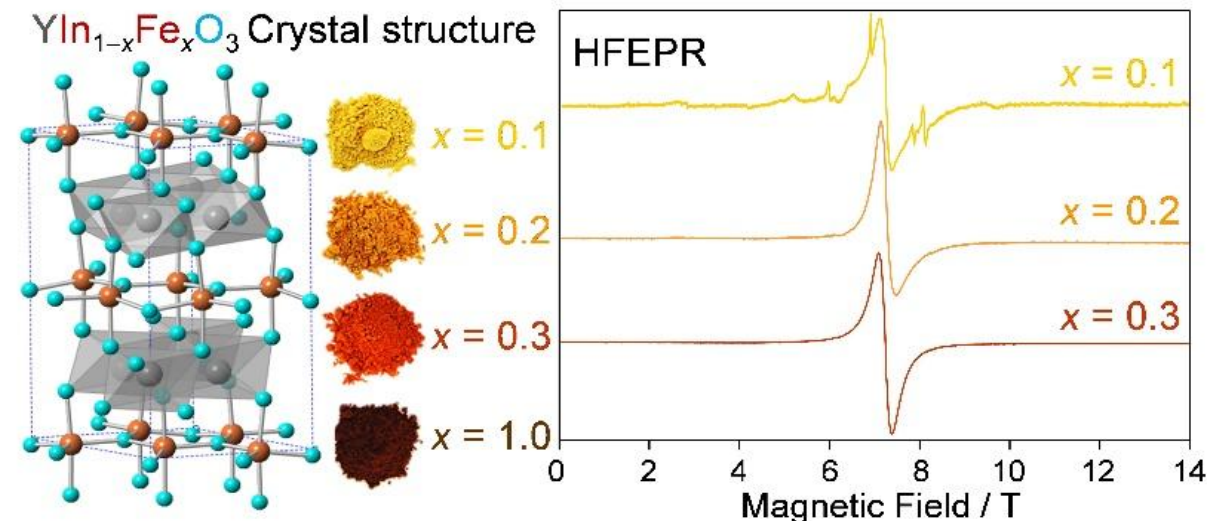


Figure 1. (Left) Crystal structure of $\text{YIn}_{1-x}\text{Fe}_x\text{O}_3$ showing layers of MO_5 ($M = \text{In}, \text{Fe}$) trigonal bipyramids (TBP) interspersed with layers of YO_6 octahedra (Y, gray spheres; In/Fe, orange spheres; O, turquoise spheres). (Center) Colors of $\text{YIn}_{1-x}\text{Fe}_x\text{O}_3$ samples varying from yellow to orange and red/brown with increasing Fe content ($x = 0$ is off-white). (Right) HFEPR spectra of $\text{YIn}_{1-x}\text{Fe}_x\text{O}_3$ for different doping levels of iron at 203.2 GHz and 10 K.

Facilities and instrumentation used: The EMR Facility at the NHMFL / Tallahassee was used including its 15/17 T superconducting magnet.

Citations: (1) Krzystek, J.; Telser, J.; Li, J.; Subramanian, M.A., Magnetic properties and electronic structure of manganese-based blue pigments: a high-frequency and -field EPR study, *Inorganic Chemistry*, 54, 9040-9045 (2015) doi.org/10.1021/acs.inorgchem.5b01306; (2) Krzystek, J.; Telser, J.; Jiang, P.; Gopakumar, G.M.; Subramanian, M.A., High-Frequency and -Field Electron Paramagnetic Resonance Studies of Yellow/Orange/Red Pigments Based on Fe³⁺ in Trigonal Bipyramidal Coordination, *Inorganic Chemistry*, 65, 12524-12531 (2026) doi.org/10.1021/acs.inorgchem.6c01324.